

2013

1. C
 2. A
 3. B
 4. D
 5. B
 6. D
 7. A
 8. B
 9. C
 10. C
11. Use $P_1/T_1 = P_2/T_2$; $P = 88.8 \text{ kPa}$, so yes, it is in the allowable range
12. $P \text{ of } \text{CO}_2 = 6.8 \times 10^3 \text{ kPa}$
 $P \text{ of } \text{H}_2 = 1.4 \times 10^4 \text{ kPa}$
13. 14. 1 L
 14. -1298 kJ
 15. 34 kJ
 16. $5.0 \times 10^1 \text{ }^\circ\text{C}$
 17. -420 kJ/mole
 18. 36 kJ
19. A) 3
B) 2
C) 100kJ/mole
D) -25
20. A) add catalyst
B) low temperature
21. Exothermic, as K increased, temperature decreased
22. a) no effect
b) increase B_2O_3
c) increase temp
d) no effect
23. a) 4.6×10^4
b) C
24. $\text{HClO}_2 \quad K_a = 1.1 \times 10^{-2}$
25. 4.74 g/L